

#### VISION

*"To incorporate technical & professional skills in Mechanical Engineers to fulfill industrial & social needs".* 

#### MISSION

- To educate, guide, and mentor the students for academic excellence
- To develop technical skills and discipline among the students as per the requirement of the industry.
- To impart ethics & social values by arranging social activities.

#### Subject Name: Thermal Engineering (313310)

Date :-

Assignment No :- 1

**Course Outcome: 303.1** 

### **Topic Name :- Fundamentals of Thermodynamics**

- 1. Differentiate between open, closed, and isolated systems with suitable examples.
- 2. Explain the concepts of extensive and intensive properties. Provide examples of each type of property.
- 3. Define work and heat from a thermodynamic perspective. How do they differ from each other?
- 4. State and explain the Zeroth Law of Thermodynamics.
- 5. State the First Law of Thermodynamics.
- 6. State the Second Law of Thermodynamics in both Kelvin-Planck and Clausius statements.
- 7. Differentiate between a heat engine, heat pump, and refrigerator.
- 8. What is the Steady Flow Energy Equation (SFEE)? Apply it to the boiler, turbine, and condenser.

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Subject Name: Thermal Engineering (313310)

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**Course Outcome: 303.2** 

Assignment No :- 2

Topic Name :- Ideal Gases and Steam Fundamentals

- 1. Explain the following ideal gas processes: isobaric, isochoric, isothermal, isentropic, and polytropic. also draw on P-V and T-S diagrams
- 2. Explain wet, dry, and superheated steam.
- 3. Define sensible heat, latent heat, total heat, specific volume, and dryness fraction.
- 4. Draw a neat sketch of the Rankine cycle on P-V and T-S diagrams and explain the processes involved in the Rankine cycle.
- 5. A gas occupies a volume of 2 liters at a pressure of 3 atm. It is heated at constant pressure until the volume becomes 4 liters. Calculate the work done by the gas.
- 6. A gas occupies a volume of 5 liters at a pressure of 4 atm. It undergoes an isothermal expansion until the pressure becomes 2 atm. Calculate the final volume.

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Assignment No :- 3

Course Outcome: 303.3

### **Topic Name :- Components of Steam Power Plant**

- 1. Define a steam power plant and explain its basic components with a neat diagram.
- 2. What is the role of a steam boiler in a steam power plant? Briefly describe the different types of boilers used in power plants.
- 3. Explain the construction and working of Lamont boiler and Benson boiler with a suitable diagram.
- 4. Explain the function and types of steam nozzles used in steam power plants.
- 5. Explain the need for compounding in steam turbines and discuss different methods of compounding
- 6. State Dalton's law of partial pressure and explain its significance in the context of steam condensers.
- 7. Explain the construction and working of a natural draft cooling tower with a suitable diagram. Discuss the advantages and disadvantages of natural draft cooling towers.

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Assignment No :- 4

Course Outcome: 303.4

### **Topic Name :- Heat Transfer & Heat Exchangers**

- 1. Define the three modes of heat transfer: conduction, convection, and radiation. Provide examples of each mode in everyday life.
- 2. Define absorptivity, transmissivity, reflectivity, and emissivity in the context of radiation heat transfer. Explain the concept of a black body and a gray body.
- 3. Explain the construction and working of a shell-and-tube heat exchanger. Discuss its applications in various industries.
- 4. Describe the construction and working of a plate-type heat exchanger. Compare its advantages and disadvantages with shell-and-tube heat exchangers.
- 5. A composite wall consists of two layers: a 10 cm thick layer of brick with a thermal conductivity of 0.8 W/mK and a 5 cm thick layer of insulation with a thermal conductivity of 0.04 W/mK. The temperature difference between the inner and outer surfaces of the wall is 60°C. Calculate the rate of heat flow per unit area through the wall.

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Subject Name: Thermal Engineering (313310)

Date :-

Assignment No :- 5

**Course Outcome: 303.5** 

### Topic Name :- Introduction to I.C. Engine & Power Cycles

- 1. Describe the Carnot cycle and its significance in thermodynamics. Represent the Carnot cycle on a P-V and T-S diagram.
- 2. Explain the Otto cycle and its applications in internal combustion engines. Represent the Otto cycle on a P-V and T-S diagram.
- 3. Differentiate between the Diesel cycle and the Otto cycle.
- 4. Define the following terms related to internal combustion engines: bore, stroke, compression ratio, mean effective pressure, and indicated thermal efficiency.
- 5. Compare two-stroke and four-stroke internal combustion engines. Explain their construction and working principles.
- 6. Classify internal combustion engines based on their fuel type, ignition system, and number of strokes. Provide examples of each type.

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